

Name the following ionic compounds:

1. NaCl _____
2. CaF₂ _____
3. FeO _____
4. Fe₂O₃ _____
5. NH₄C₂H₃O₂ _____
6. Na₂CrO₄ _____

Give the formula for the following ionic compounds.

7. lithium bromide _____
8. chromium (II) oxide _____
9. tin (IV) oxide _____
10. calcium phosphate _____
11. calcium phosphide _____
12. calcium phosphite _____

Name the following ionic compounds:

1. NaCl sodium chloride
2. CaF₂ calcium fluoride
3. FeO iron (II) oxide
4. Fe₂O₃ iron (III) oxide
5. NH₄C₂H₃O₂ ammonium acetate
6. Na₂CrO₄ sodium chromate

Give the formula for the following ionic compounds.

7. lithium bromide LiBr
8. chromium (II) oxide Cr₂O₃
9. tin (IV) oxide SnO₂
10. calcium phosphate Ca₃(PO₄)₂
11. calcium phosphide Ca₂P₂
12. calcium phosphite Ca₂(PO₃)₂

