## General Chemistry

Mr. MacGillivray
Quiz \#31:
Molarity
Solve the following problems. Show all work.

1. Determine the concentration of a solution in which 1.50 mol of NaCl is dissolved in enough water to make 12.0 L of solution.
2. Determine the concentration of a solution in which 1.50 g of NaCl is dissolved in enough water to make 12.0 L of solution.
3. A chemistry student needs 50.0 ml of $0.035 \mathrm{M} \mathrm{NaCl}(\mathrm{aq})$ in order to carry out a lab experiment. However, the only solution available in the lab is $0.100 \mathrm{M} \mathrm{NaCl}(\mathrm{aq})$. Explain how she would make the dilute solution that is needed for her experiment.
