## Practice Problems: Measurement

1. Express 549000000 in scientific notation.
[A] $5.49 \times 10^{-8}$
[B] $5.49 \times 10^{8}$
[C] $549 \times 10^{8}$
[D] $549 \times 10^{6}$
[E] $54.9 \times 10^{-7}$
2. Express 506100 in scientific notation.
[A] $5.06100 \times 10^{5}$
[B] $5.1 \times 10^{5}$
[C] $5 \times 10^{5}$
[D] $5.061 \times 10^{5}$
[E] $51 \times 10^{5}$
3. Write 4,251 in standard scientific notation.
[A] 4.251
[B] $4.251 \times 1000$
[C] $42.52 \times 10^{2}$
[D] 4,251
[E] $4.251 \times 10^{3}$
4. The number 0.000402 expressed in exponential notation is
[A] $4.02 \times 10^{3}$
[B] $4.02 \times 10^{-3}$
[C] $4.02 \times 10^{-4}$
[D] $4.02 \times 10^{-4}$
[E] $4.02 \times 10^{-2}$
5. One kilogram contains this many grams.
[A] 1000
[B] 10
[C] 1/100
[D] $1 / 1000$
[E] 100
6. How many milliliters are in 0.020 L ?
[A] 200 mL
[B] 0.20 mL
[C] 2.0 mL
[D] 20. mL
[E] 5.0 mL
7. The measurement $4.3 \times 10^{3} \mathrm{~g}$ also could be written as
[A] 4.3 dg
[B] 4.3 kg
[C] 4.3 pg
[D] 4.3 mg
[E] 4.3 g
8. How many millimeters are in 251 centimeters?
[A] $2.51 \times 10^{3} \mathrm{~mm}$
[B] 2.51 mm
[C] $2.51 \times 10^{2} \mathrm{~mm}$
[D] $2.51 \times 10^{-2} \mathrm{~mm}$
[E] $2.51 \times 10^{1} \mathrm{~mm}$
9. Convert: $1 \mathrm{~cm}=$ $\qquad$ mm.
10. Convert: $4.96 \mathrm{~kg}=$ $\qquad$ mg.
11. Convert: $683 \mathrm{~mm}=$ $\qquad$ cm.
12. Convert: $25 \mathrm{~mL}=$ $\qquad$ L.
13. One millisecond is equal to how many seconds?
[A] $10^{-3} \mathrm{~s}$
[B] $10^{6} \mathrm{~s}$
[C] $10^{-6} \mathrm{~s}$
[D] $10^{3} \mathrm{~s}$
[E] 1 s
14. The fundamental unit of mass in the metric system is the
[A] milliliter
[B] kilometer
[C] centimeter
[D] meter
[E] gram
15. A cubic centimeter $\left(\mathrm{cm}^{3}\right)$ is equivalent to what other metric volume unit?
[A] millimeter
[B] liter
[C] centimeter
[D] deciliter
[E] milliliter
16. The number of millimeters in 0.101 meter is
[A] $9.90 \times 10^{3} \mathrm{~mm}$
[B] $1.01 \times 10^{-3} \mathrm{~mm}$
[C] $1.01 \times 10^{-4} \mathrm{~mm}$
[D] $1.01 \times 10^{4} \mathrm{~mm}$
[E] $1.01 \times 10^{2} \mathrm{~mm}$
17. The number of cubic centimeters $\left(\mathrm{cm}^{3}\right)$ in 43.0 mL is
[A] $0.0430 \mathrm{~cm}^{3}$
[B] $43.0 \mathrm{~cm}^{3}$
[C] $4.30 \mathrm{~cm}^{3}$
[D] none of these
18. A student finds that the weight of an empty beaker is 12.024 g . She places a solid in the beaker to give a combined mass of 12.108 g . To how many significant figures is the mass of the solid known?
[A] 4
[B] 2
[C] 3
[D] 5
[E] 1
19. How many significant figures are in the number $4.00700 \times 10^{13}$ ?
[A] none of these
[B] 2
[C] 4
[D] 6
[E] 5
20. Convert 258 L to milliliters.
[A] $2.58 \times 10^{3} \mathrm{~mL}$
[B] $2.58 \times 10^{5} \mathrm{~mL}$
[C] 0.258 mL
[D] 258 mL
[E] 2.58 mL
21. Convert 561097 mm to kilometers.
[A] 561.097 km
[B] $5.61097 \times 10^{11} \mathrm{~km}$
[C] 5.61097 km
[D] 0.561097 km
[E] 5610.97 km
